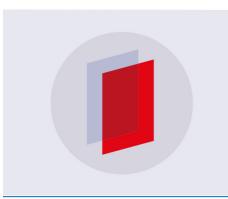
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Strongly inhomogeneous distribution of spectral properties of silicon-vacancy color centers in nanodiamonds

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Keywords: color centers, diamond, nanodiamond, optical emission, single photons, density functional theory calculations

Abstract

The silicon-vacancy (SiV) color center in diamond is a solid-state single photon emitter and spin quantum bit suited as a component in quantum devices. Here, we show that SiV centers in nanodiamonds exhibit a strongly inhomogeneous distribution with regard to the center wavelengths and linewidths of the zero-phonon-line (ZPL) emission at room temperature. We find that the SiV centers separate in two clusters: one group exhibits ZPLs with center wavelengths within a narrow range \approx 730–742 nm and broad linewidths between 5 and 17 nm, whereas the second group comprises a very broad distribution of center wavelengths between 715 and 835 nm, but narrow linewidths from below 1 up to 4 nm. Supported by *ab initio* Kohn–Sham density functional theory calculations we show that the ZPL shifts of the first group are consistently explained by strain in the diamond lattice. Further, we suggest, that the second group showing the strongly inhomogeneous distribution of center wavelengths might be comprised of a new class of silicon-related defects. Whereas single photon emission is demonstrated for defect centers of both clusters, we show that emitters from different clusters show different spectroscopic features such as variations of the phonon sideband spectra and different blinking dynamics.

1. Introduction

Recently, the negatively charged silicon-vacancy center (SiV center) has received growing interest in the fields of quantum communication and quantum information due to its favorable spectral properties and optically accessible spin states [1–4]. In particular, SiV centers in low strain bulk diamond have been shown to exhibit close to Fourier-limited linewidths [5–7] and emission of indistinguishable photons [8] from two distinct emitters, a prerequisite for many applications in quantum technologies [9]. As the spin coherence time of SiV centers in bulk diamond is limited by phonon-induced decoherence processes down to liquid-Helium temperatures [1, 3, 10], diamond nanostructures or nanodiamonds might provide an advantage based on phonon confinement effects modifying the phonon density of states. Additionally, some applications require diamonds of the size of a few ten nanometers. Examples include applications as fluorescence markers [11, 12] or the implementation of SiV centers in photonic structures such as microcavities [13, 14] or optical antennas [15, 16].

In light of such applications, nanodiamonds hosting color centers provide a significant advantage: Suitable SiV centers can be spectroscopically preselected according to desired properties and can then be relocated to be used in the target structure using pick-and-place techniques [17]. Previous research demonstrated the production of nanodiamonds including SiV centers produced by chemical vapor deposition (CVD) [18],

Table 1. Overview of the investigated nanodiamond samples. The columns indicate sample names, the mean diameter of the nanodiamonds, the SiV center incorporation method, and the post-processing treatment(s) of the samples.

Sample name	Diameter	Sivincorporation	Post-processing
insitu50	50 nm	In situ	Series of individual samples with combinations of annealing and oxidation
insitu70	70 nm	In situ	Series of individual samples with combinations of annealing and oxidation
insitu70n	70 nm	In situ	No post-processing subset of insitu70
insitu70o	70 nm	In situ	oxidized in air at 450 °C subset of insitu70
insitu100	100 nm	In situ	Series of individual samples with combinations of annealing and oxidation
insitu100ao	100 nm	In situ	Annealed in vacuum at 900 °C, consecutively oxidized in air at 450 °C subset of insitu100
implanted250ao	250 nm	<i>I</i> mplanted	Annealed in vacuum at 900 $^{\circ}\mathrm{C}$, consecutively oxidized in air at 450 $^{\circ}\mathrm{C}$

high-pressure, high-temperature [19] and bead-assisted sonic disintegration (BASD) [20] processes. Although several studies [21, 22] have indicated nearly Fourier-limited linewidths of SiV centers in nanodiamonds, in general the spectral properties of SiV centers are strongly dependent on strain in the diamond lattice [23–25].

In this study we investigate the properties of SiV centers in nanodiamonds of sizes of 50–100 nm produced in a wet-milling process in a vibrational mill. Starting material was a CVD grown diamond film with *in situ* incorporated SiV defects. The advantage of wet-milled nanodiamonds with *in situ* incorporated SiV centers lies in its high production rate of nanodiamonds, yielding a perfect candidate for preselection of spectral properties and consecutive implementation in target applications. To verify that all observed emitters are silicon-related we additionally investigate nanodiamonds where silicon is implanted into initially silicon-free CVD diamond material.

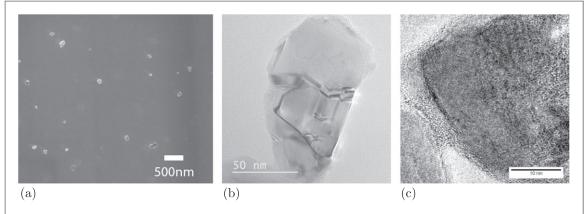
The fluorescence spectra of the SiV centers show that both the center wavelength of the zero-phonon-line (ZPL) as well as the linewidth of the ZPL vary strongly among different diamonds. Our measurements show a strong correlation between the center wavelength of the ZPL and the corresponding linewidths, resulting in a previously unreported bimodal distribution. We find single photon emission from these SiV centers over the whole range of ZPL positions and linewidths, although with a large variation in photon emission rate and fluorescence stability.

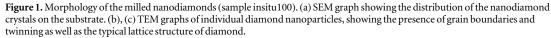
2. Methods

2.1. Nanodiamond production

In the following, we describe the processes used to produce nanodiamond samples of interest. We deploy two different methods of incorporating SiV centers into wet-milled nanodiamonds: incorporation during (*in situ*), and implanting after the diamond growth process. The obtained samples are prefixed *in situ* and *implanted* respectively, see table 1 for an overview of the available samples.

The starting material for the wet-milled nanodiamonds was a nanocrystalline diamond film [26], directly grown on a silicon wafer using CVD. A microwave hydrogen plasma containing 1% methane was used to grow on purified 5 nm nanodiamond seeds (produced by PlasmaChem). To induce in situ SiV center creation, sacrificial silicon pieces are situated in the growth chamber. During diamond growth the silicon pieces are etched by the plasma and individual silicon atoms are incorporated into the diamond lattice. We verified successful creation of SiV centers in the diamond films by optical spectroscopy revealing a homogeneous distribution over the sample (corresponding to an areal density $\gg 10^8$ cm⁻²). Using a wet-milling process, the diamond film after removal of the substrate is milled with steel beads in a vibrational mill. The resulting particle suspension is fractionated using centrifugation, yielding 3 sets of diamond particles of sizes of about 50; 70; 100 nm in average diameter (figure 1(a)), as determined with dynamic light scattering. Transmission electron microscopy (TEM) graphs of the milled diamond particles show that the nanodiamonds are polycrystalline and exhibit typical single-crystal sizes of a few tens of nanometers. In figures 1(b) and (c) TEM images of a typical nanodiamond are shown. Within the nanodiamond, several sharp lines are visible. These lines are edges of crystal boundaries and grain boundaries, introducing strain in the diamond lattice. We remark at this point that some studies suggest the possibility that SiV centers are created with a higher probability at grain boundaries and morphological defects than within the core of the crystal [27, 28]. The high amount of debris from milling beads is removed for the most part by extensive acid treatment and the absence of debris shown by spectroscopic characterization. We also explored milling nanodiamonds with silicon nitride beads, and found that the choice of bead material did not cause any noticeable spectroscopic difference. The aqueous solution containing the nanodiamonds is drop-cast onto an iridium film on a silicon substrate. The 130 nm iridium film is grown onto a buffer layer of yttria-stabilized zirconia, which in turn is grown onto a silicon wafer. The iridium surface has the





advantage that it acts as a broad-band optical antenna, uniformly enhancing the collection efficiency of fluorescence light over the entire SiV center emission range [29]. Prior to drop-casting, the substrate was cleaned using Piranha solution (50% sulfuric acid H_2SO_4 , 50% hydrogen peroxide H_2O_2). This enhances surface hydrophilicity, leading to a homogeneous distribution of the diamond particles on the substrate. Post-processing treatment is comprised of annealing in vacuum at 900 °C, consecutive oxidation in air at a temperature of 450 °C or a combination thereof. The duration for either treatment method was 3–6 h.

As mentioned before, we also investigated nanodiamonds with SiV centers implanted after completed diamond growth. As starting material we used a polycrystalline diamond film (Element Six, electronic grade) for which we verified that it did not contain SiV centers initially. In bulk material, the implantation causes the SiV centers to form in a specific depth dependent on the implantation energy, leaving most of the diamond vacant of SiV centers. As a consequence, a significant portion of nanodiamonds milled from such a bulk material would not host any SiV centers. To obtain diamond particles with a homogeneous distribution of SiV centers, the following steps were taken: first, the diamond film was milled to diamond particles of sizes on the order of a few micrometers. In a second step, these microdiamonds were densely spin-coated onto iridium substrates and implanted with silicon (implantation energy 900 keV; implantation depth 500 ± 50 nm; fluence 10^{11} cm⁻²; yield >1%, resulting in an SiV areal density of >10⁹ cm⁻²). To eliminate damage from the implantation process, the diamonds were annealed in vacuum at 900 °C and subsequently oxidized in air at 450 °C for 3 h each. At this stage, we verified successful creation of SiV centers via optical spectroscopy. Finally, the micrometer sized diamond particles were milled to a size of 250 nm.

2.2. Experimental setup

We obtained photoluminescence (PL) spectra and photon statistics of the samples using a home-built confocal microscope. For excitation we use a continuous wave diode laser (Schäfter-Kirchhoff, 58FCM) at 660 nm. The excitation laser is focused onto the sample via a microscope objective with a numerical aperture of 0.8 (Olympus, LMPlanFLN 100x), which also collects the fluorescence light emitted from the sample. Both the laser light reflected by the sample and the fluorescence light pass through a glass plate used to couple the excitation laser into the microscope objective. The residual laser light is then filtered out by two 710 nm longpass filters. The emission light is coupled into a single mode fiber which serves as pinhole in the confocal setup. The emission is either guided to a grating spectrometer or to two single photon detectors (PicoQuant, tau-SPAD-100) used in a HBT configuration to measure photon statistics. In front of the avalanche photo diodes bandpass filters select the spectral window in which the investigated color centers emit photons. These filters are chosen according to each individual emitter: due to strong shifts in the ZPL wavelength, we use different bandpass filters in front of the APDs to suppress background fluorescence from the diamond material and effectively select the luminescence of the ZPL. All experiments were performed at room temperature.

3. Results

In the following, we present our findings regarding diamond crystal characteristics and spectroscopic measurement of SiV centers. Unless explicitly stated, we rely on measurements of milled nanodiamonds containing *in situ* incorporated SiV centers, i.e. samples insitu50, insitu70, and insitu100 as listed in table 1.

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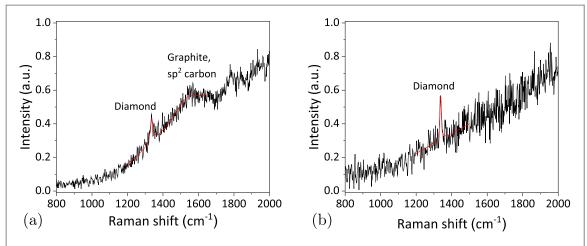


Figure 2. Raman measurements, black: data, red: fit. (a) Raman measurement before oxidation, sample insitu70. The diamond Raman peak is situated at 1338 cm⁻¹. The broad feature around 1580 cm⁻¹ corresponds to the graphite G-band. (b) Raman measurement after oxidation, sample insitu700. The G-band has vanished, indicating removal of the majority of graphite and amorphous sp² hybridized carbon.

3.1. Diamond crystal quality

We aim to perform spectroscopic measurements of single SiV centers in wet-milled nanodiamonds. To this end, we focus on producing pristine wet-milled diamond nanoparticles containing a single SiV center each. Raman measurements of the nanodiamonds allow us to identify issues with surface contamination, defects of the diamond lattice, and strain in the diamond lattice [30–32]. Surface contamination like graphite and amorphous sp² hybridized carbon manifest themselves as additional peaks in the Raman spectrum. Strain in the diamond lattice broadens the first order Raman peak and causes it to shift to higher or smaller wavenumbers. Similarly, high concentrations of lattice defects cause additional peaks, a broadening of the first order Raman peak and a shift towards smaller wavenumbers. The size of single nanodiamonds is on the order of tens of nanometers, thus low signal intensities can become an issue. To overcome this problem we pursue two different approaches to perform Raman measurements:

- (a) Nanodiamond clusters: collective measurements are carried out at several areas on the sample insitu70. Since this sample is densely covered with nanodiamonds, collective measurements of clusters of nanodiamonds (figure 2(a)) achieves higher signal intensities.
- (b) Large nanodiamonds: raman measurements are carried out on the implanted sample implanted250ao. For this sample, diamond particles are large enough to yield sufficient intensities on single nanodiamonds.

For all Raman measurements a 532 nm continuous wave diode laser was used for excitation.

3.1.1. Surface contamination

We test the impact of oxidation treatment as described in section 2 on surface contamination. Figure 2(a) shows a measured Raman spectrum of a sample without oxidation treatment (insitu70n). To verify reproducibility, the measurement is performed on three different spots of the sample. The narrow peak in figure 2(a) corresponds to the first order diamond Raman peak and will be further analyzed in below. The spectrum also shows a broad peak with a Raman shift of about ($1582 \pm 5 \text{ cm}^{-1}$. This shift corresponds to the G-band due to amorphous sp² hybridized carbon atoms and graphite. The exact G-band position and linewidth is sensitive to parameters such as the clustering of the sp² phase, bond-length and bond-angle disorder, presence of sp² rings or chains, and the sp²/sp³ ratio [33]. The nanodiamond Raman spectra are considerably modified after oxidation in air at 450 °C. To verify this, we perform Raman measurements on three different spots of a sample produced in the same process as the above mentioned, which is additionally oxidized (insitu700). While the G-band peak is present in every measurement performed on a non-oxidized sample, it is not present in any of the oxidized samples (figure 2(b)), indicating successful removal of a majority of sp² hybridized carbon and surface graphite.

3.1.2. Defects

Several effects impact the first order diamond Raman line: (i) defects in the diamond lattice, (ii) hydrostatic pressure, (iii) uniaxial or more complicated stress configurations. In the measurement on nanodiamond clusters the width of the diamond Raman peak of sample insitu70 varies between 15 and 30 cm⁻¹ without oxidation

treatment, but is only 9–11 cm⁻¹ after the oxidation process. A possible reason for this change of the width is improved crystal quality [20, 31]. In the measurement on large nanodiamonds we measured a Raman line at (1308 \pm 5) cm⁻¹ (denoted line R1) which exhibits a broad linewidth of (25 \pm 5) cm⁻¹. One plausible explanation for both the position and the linewidth of the Raman line are defects in the diamond lattice [31], originating from the implantation process.

3.1.3. Strain

We investigated how strain in the diamond lattice, as induced during CVD growth of the nanocrystalline starting material, manifests itself in both measurements on nanodiamond clusters and on large nanodiamonds. In the Raman measurement on nanodiamond clusters, the position of the diamond Raman peak is the same for oxidized (insitu700) and non-oxidized (insitu70n) samples, indicating that oxidation does not affect strain in the diamond. However, the Raman shift of both non-oxidized and oxidized samples amounts to (1338 ± 5) cm⁻¹, as compared to the literature value of 1 332.5 cm⁻¹ of pristine diamond [30] (given uncertainties are governed by spectrometer resolution). This shift indicates the presence of strain in the diamond particles.

Performing the Raman measurement on large nanodiamonds we find diamond Raman lines between $(1308 \pm 5) \text{ cm}^{-1}$ (line R1) and $(1348 \pm 5 \text{ cm}^{-1}$ (line R2), indicating a broad distribution of strain among the individual diamond particles (uncertainties governed by spectrometer resolution). Only line R1 could be explained with a high defect concentration in the diamond lattice due to its shift to smaller wavelengths. However, a more consistent model which explains all occurring shifts is the presence of strain/stress in the diamond nanoparticles. The Raman shift $\Delta \tilde{\nu}$ in the presence of compressive and tensile stress is given by [34, 35]: $\Delta \tilde{\nu} = p/0.34$, where the Raman shift $\Delta \tilde{\nu}$ is given in cm⁻¹ and the stress *p* in GPa. The calculation yields a pressure range from $(-8.33 \pm 1.7 \text{ GPa tensile stress to } 5.27 \pm 1.7 \text{ GPa compressive stress. Whereas under hydrostatic pressure the triply degenerate first order Raman peak remains degenerate, under uniaxial and more complex stress configurations (biaxial stress, shear stress etc) mode splitting occurs [31]. As mentioned above, we observe broad linewidths up to <math>(25 \pm 5) \text{ cm}^{-1}$. The broad Raman linewidths may be attributed to uniaxial strain where mode splitting manifests itself in a broadening of the peak due to limited spectrometer resolution.

3.2. PL spectra

To identify nanodiamonds containing SiV centers, we performed confocal scans of the samples. To reduce bias in the measurements, not only the brightest spots of the confocal scans are investigated, but also those which barely exceed background fluorescence. SiV centers are further investigated by measuring PL spectra, single photon statistics and photostability. The typical luminescence spectrum of an SiV center is composed of a prominent ZPL and weak sidebands. Investigations of both are reported independently in the following paragraphs.

3.2.1. Zero-phonon-line

3.2.1.1. ZPL distribution

The center wavelength and the linewidth of the ZPL of SiV luminescence spectra for samples insitu50, insitu70, and insitu100 are determined by fitting a Lorentzian fit to the ZPL. Both spectra from single and multiple SiV centers are taken into account. In figure 3 the linewidth for each measured ZPL is plotted against its center wavelength. What immediately strikes the eye is a pattern that to our knowledge has not been reported to date: The observed ZPLs partition into two groups, here denoted as horizontal cluster (group H) and a vertical cluster (group V). The two distributions are separated by a gap, i.e. a region with a pronounced lack of data points. Single emitters are found both in group H and group V, marked as red triangles in figure 3. Further details on single emitters are given in section 3.3.

The two groups are defined by their characteristic center wavelengths and linewidths: in group H very prominent ZPL peaks are found showing linewidths in the range of 1–5 nm and center wavelengths in the range of 715–835 nm. Figure 4(a) shows a representative spectrum of a single emitter in group H (denoted emitter H1), exhibiting a ZPL linewidth of 1.4 nm and a center wavelength of 726.5 nm. In contrast, in group V the spectra exhibit broader ZPL linewidths of approximately 5 nm up to 18 nm. Their ZPL center wavelengths, however, are distributed within a narrow range of 730–741 nm. Figure 4(b) shows a spectrum of a single emitter of group V (denoted emitter V1) with a ZPL linewidth of 6.4 nm and a center wavelength of 740.8 nm. For comparison, the room temperature ZPL of SiV centers in unstrained bulk diamond (see e.g. [2]) exhibits a linewidth of 4nm to 5 nm and a center wavelength of 737.2 nm marked with a black cross in figure 3 [6, 36].

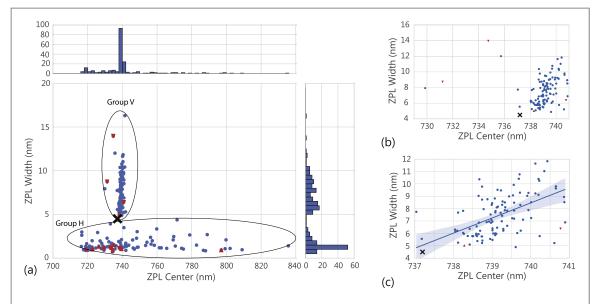
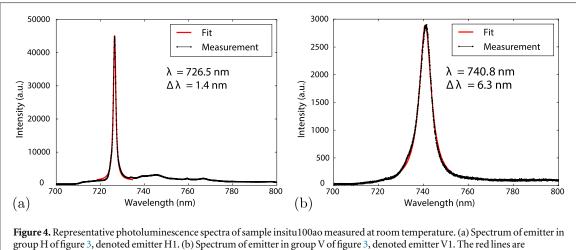


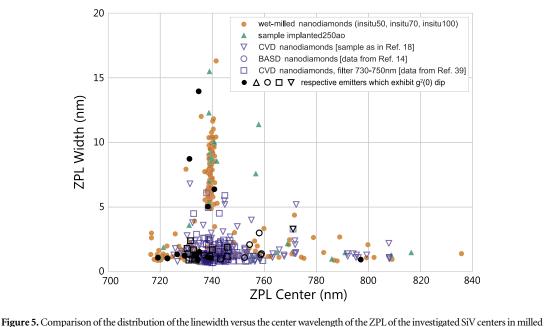
Figure 3. (a) Distribution of the ZPL linewidth versus the center wavelength of the investigated SiV centers in milled nanodiamonds containing *in situ* incorporated SiV centers for samples insitu50, insitu70, insitu100. The data separates into a horizontal (group H) and a vertical (group V) cluster. The bold black cross marks the position of an ideal SiV center in unstrained bulk diamond [6]. The red triangles indicate emitters with an antibunching dip, $g^{(2)}(0) < 0.5$. Upwards pointing triangles represent blinking emitters (fluorescence intermittency), while triangles pointing down represent non-blinking emitters (see section 3.4). (b) A zoom into group V. While many data points exhibit higher center wavelengths (i.e. a redshift) than the ideal SiV center in bulk, only few exhibit shorter center wavelengths (i.e. a blueshift). (c) Zooming further into group V, a clear trend of broader ZPL linewidth for larger ZPL center shifts is visible. The line is a linear regression to all datapoints between 737 and 741 nm which exhibit a linewidth larger than 4 nm. For the linear regression the least trimmed sum of squares method was used. The shaded area indicates the 95% confidence interval of the regression determined by bootstrapping.

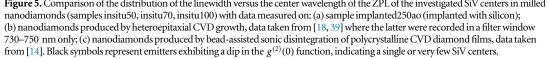


Lorentzian fits to the peaks.

3.2.1.2. Debye-Waller factor

To determine how much the ZPLs contribute to the total observed emission of emitter H1 and emitter V1, we determine the Debye–Waller factor defined as $DW = I_{ZPL}/I_{TOT}$ where I_{ZPL} and I_{TOT} are intensities in ZPL and total spectrum, respectively. To this end, background-corrected (horizontal baseline) PL spectra were used. The Debye–Waller factor for emitter H1 amounts to 0.81 ± 0.01 (given uncertainty due to fit). This Debye–Waller factor corresponds to a Huang-Rhys factor $S = -\ln(DW)$ [37] of 0.21 ± 0.01 , which is in good agreement with the values reported in [18]. The error is mainly due to background corrections. When zooming in onto the spectrum of emitter V1 we do not find distinct sidebands peaks, i.e. almost all emission for this emitter is contained within the ZPL. Considering resolution limits of the spectrometer, dark counts and fluorescence background, we evaluate the Debye–Waller factor to be larger than 0.97 which is the largest Debye–Waller factor among all investigated SiV centers. The two emitters H1 and V1 discussed here are typical representatives of their group (H and V, respectively), showing typical emission spectra as presented above. The corresponding Debye–Waller factors, however, are not necessarily representative for the two groups. Rather, from investigations of many emitters we find no systematic difference of the Debye–Waller factor between groups H





and V. The Debye–Waller factor strongly depends on the nature of the electron–phonon coupling [38]. Strain in the nanodiamonds distorts the SiV center symmetry and thus the electron–phonon coupling, giving rise to varying Debye–Waller factors (see also discussion of strain effects below).

3.2.1.3. Comparison to earlier results

To provide context for the novel findings presented in figure 3, we compare our results to various earlier findings. Furthermore, we discuss an additional comparison to an investigated control sample fabricated using silicon implantation. The results are presented in figure 5.

Samples for which previous data has been taken are:

- (i) nanodiamonds produced by BASD of polycrystalline CVD diamond films [20] (open blue circles in figure 5; data taken from [14])
- (ii) nanodiamonds produced by heteroepitaxial CVD growth on Ir substrates [18] with *in situ* incorporated SiV centers; measured in a spectral filter window of 730–750 nm (blue squares in figure 5; data reused from [39] with permission)
- (iii) nanodiamonds as in (ii) (blue downwards pointing triangles in figure 5; spectroscopic measurement performed with setup described in section 2, no filter window)

All previous data from different nanodiamond material fit nicely with the ZPL distribution presented in figure 3, confirming the findings discussed above. We verify that the observed luminescent centers are indeed silicon-related defects by performing control experiments with silicon implanted samples (sample implanted250ao). By doing so we rule out the possibility that the two clusters in the distribution are a result of artifacts. Such artifacts include other elements incorporated into the nanodiamonds during the growth process: residue from previous processes performed in the diamond growth chamber or material from chamber parts may be incorporated during nanodiamond growth. Figure 5 shows that the implanted SiV centers cover the same spectral range from around 720–820 nm as the *in situ* incorporated centers. This correlation provides strong evidence for the silicon related origin of the defects.

3.2.2. Strain shift of the ZPL

3.2.2.1. Discussion of ZPL strain shift

To provide a theoretical interpretation, the ZPL center wavelength shift is investigated in further detail and compared to results from density functional theory (DFT) calculations: zooming in to group V (figure 3(b)) it

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becomes clear that only six of the measured data points in group V are situated at a shorter center wavelength than the point attributed to an ideal SiV center in unstrained bulk material, the shortest wavelength ZPL being situated at 729.9 nm. At the same time, much more data exhibit a center wavelength red-shifted to the ideal SiV center. This asymmetry suggests that a red-shift of the ZPL of an SiV center is significantly more likely than a blueshift. Several mechanisms contribute to the center wavelength shift, predominantly hydrostatic and uniaxial strain. As discussed in section 3.1, we estimate the stress in our nanodiamond sample from Raman measurements to be on the order of -8.33 to 5.27 GPa. In the following, we first discuss the stress/strain shift mechanisms for the SiV center before we compare theoretically derived strain shift coefficients to the mentioned range of ZPL shifts.

3.2.2.2. Theoretical model

To gain insight into the strain mechanism for the SiV centers in diamond, we perform *ab initio* Kohn–Sham DFT calculations on the strain ZPL shift coupling parameters. The unstrained model of the negatively charged silicon vacancy defect (SiV⁻) in bulk diamond is constructed starting from a 512 atom pristine diamond simple cubic supercell within the Γ point approximation. The Γ point sampling of the Brillouin zone has proven to be adequate for defects in 512-atom supercells [40, 41] providing a sufficiently converged charge density. The SiV⁻ defect has $S = \frac{1}{2}$ and it is found to have D_{3d} symmetry with an axis oriented along $\langle 111 \rangle$. Standard projector augmented-wave formalism together with plane waves are applied, as implemented in the Vienna *Ab initio* Simulation Package code [42–45]. The geometry optimization is carried out within the constructed supercell by using the Perdew–Burke–Ernzerhof (PBE) [46] DFT functional. A 420 eV cutoff is applied for the wave function expansion and a 1260 eV cutoff for the charge density. The geometry of the defect is optimized until the forces were lower than 10^{-6} eV Å. The D_{3d} symmetry is preserved for both the ground state and the excited state after relaxation.

The ground state of the defect is found to have electronic configuration $e_u^4 e_g^3 ({}^2E_g)$ while the excited state is modeled by promoting one electron from the e_u to the e_g level and presents electronic configuration $e_u^3 e_g^4 ({}^2E_u)$. Both these states are dynamic Jahn–Teller systems [47, 48]. The optical signal of the defect (ZPL) can be calculated as the lowest excitation energy by the constraint DFT approach (CDFT) [49]. According to CDFT one electron is promoted from the ground state to a higher level leaving a hole behind. The interaction between the electron and the hole is included in the procedure. The ZPL energies were obtained by taking the total energies of the optimized geometries in the ground and excited state.

The strain on the defect structure is simulated by applying a compression to the supercell along a well defined direction. The strained supercells are obtained by compressions along $\langle 100 \rangle$, $\langle 110 \rangle$ and $\langle 111 \rangle$. We also study the configuration produced by a hydrostatic pressure, which consists in subjecting the cell to the same compression along the three directions. After introducing the strain along the directions, the ZPL energies were calculated for each strained supercell. Finally, we obtained data points on the calculated ZPL energies versus the applied strain. These ZPL energies correspond to the optical transition between the lower branch levels of the ${}^{2}E_{u}$ and ${}^{2}E_{g}$ doublets. We note that additional calculations were performed for the $\langle 100 \rangle$ and $\langle 110 \rangle$ strained supercells by using the screened, range-separated, non-local hybrid density functional of Heyd–Scuseria–Ernzerhof (HSE06) [50, 51] and we found good agreement with the PBE values.

Nudged elastic band calculations [52] by HSE06 were performed in order to calculate the energy barriers in the ground state in strained supercells. The barrier energies between the C_{2h} configurations stayed under 10.0 meV which implies small change in the adiabatic potential energy surface around the D_{3d} symmetry upon the applied strain. As a consequence, the Ham reduction factor in strained SiV center will minutely change with respect to that of unstrained SiV center [38]. This suggests that the observed ZPL shifts upon stress are directly strain related, and the contribution of the change of the effective spin-orbit to the ZPL shifts is minor.

3.2.2.3. Comparison of theoretical strain shifts and experimental ZPL distributions

The data points in figure 6 show the ZPL center wavelength shifts calculated with the method outlined above. For comparison with experimentally determined stress data the strain values of the theoretical calculation were converted into stress assuming a simplified model where diamond is approximated as isotropic linear elastic material. In this case stress σ is related to strain ε via Young's modulus $E:\sigma = E\varepsilon$. This assumption is pragmatic as we do not know the orientation of stress in the nanodiamonds from the Raman measurements but only its modulus. The values of *E* vary considerably among different diamond materials [53] but even nanocrystalline diamond may obtain a large Young's modulus $E \ge 1000$ GPa [54]. As an average value, taking into account the nanodiamond size used in our investigations, we assume $E = 1100 \pm 100$ GPa [53]. We observe that the calculated data points for the stress shift show a linear variation for a large range of stress (up to \approx 5.5 GPa), comparable to the stress found in the nanodiamonds. We thus use linear fit functions to yield stress shift coefficients which are on the order of 1 nm GPa⁻¹ \approx 550 GHz GPa⁻¹. Stress/strain susceptibilities of the SiV

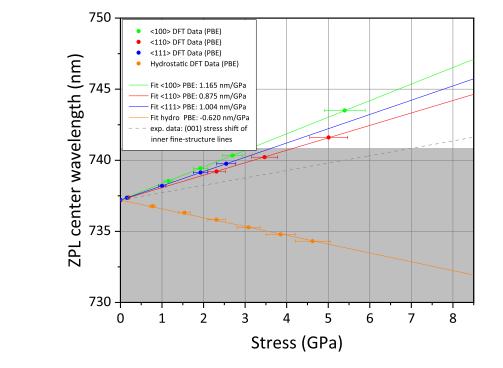


Figure 6. Calculations of the wavelength of the SiV center ZPL in dependence of pressure. Markers: DFT calculated pressure with PBE and HSE functionals; lines: linear fits to the calculated points. Hydrostatic-type pressure causes a moderate blue shift whereas uniaxial strain causes larger redshift with different magnitudes depending on the direction of the strain. Gray line: experimental stress shift data for (001) uniaxial stress and inner fine-structure line of the SiV center. Gray area: range of ZPL center wavelengths of group V.

center fine structure lines have also been determined experimentally before [55–57]: in [55] the stress shift coefficient for the SiV center under uniaxial stress of up to 0.5 GPa in $\langle 001 \rangle$ direction were measured. These measurements were performed at 4 K and reveal the shifts of the SiV center fine structure lines [56]: the outer lines of this fine structure shift with about 4 nm GPa⁻¹ (2.23 THz GPa⁻¹, not shown here), whereas the inner lines shift with only 0.52 nm GPa^{-1} (292 GHz GPa^{-1} , shown as gray dashed line in figure 6). The room temperature spectrum, however, is mostly governed by the inner line 'C' of the spectrum [6], i.e. the line with second highest wavelength, corresponding to the optical transition between the lower branch levels of the ${}^{2}E_{u}$ and ${}^{2}E_{q}$ doublets as used in the DFT calculations. Furthermore, strain susceptibilities were determined for SiV centers (at 4K) placed in a voltage-deflected diamond cantilever where strain up to 1.5×10^{-4} (<0.2 GPa) could be applied [57]. For line 'C' of the fine structure a shift of 1 PHz/strain corresponding to 900 GHz GPa⁻¹ was found. Thus, the stress shift coefficients from the DFT calculations presented here agree very well with experimentally determined susceptibilities. Most importantly, the calculated and measured stress shift coefficients together with the typical stress values in the nanodiamonds as observed from the Raman measurements explain very well the range of measured red-shifted ZPLs of group V (gray area in figure 6). We thus interpret the ZPL shifts of group V as originating from level shifts due to uniaxial strain. Furthermore, the calculated ZPL shifts due to hydrostatic pressure coincide well with the range of the blue shifted ZPLs that we observe in group V. The fact that we see only few blue-shifted ZPLs might be due to the reason that pure hydrostatic pressure is rarely observed and overlayed by uniaxial stress in the nanodiamonds. On the other hand, the measured ZPL shifts in group H would require stress up to the order of 90-200 GPa to be explained by the stress shift coefficients given above. However, usual stress values in CVD diamond amount to a few GPa only. Thus the broad distribution of defect center ZPL wavelengths cannot solely be explained by strain in the diamond. Together with the observed existence of a gap between groups H and V this points towards a separate fluorescent defect complex containing silicon, e.g. the association of SiV centers with a further nearby defect, such as a vacancy, or a modified SiV complex such as SiV:H [58].

Zooming in to group V, another trend becomes visible (figure 3(c)): with increasing ZPL center wavelength, the linewidth becomes broader. As discussed above, a red-shift of the ZPL is linked to increasing uniaxial strain. Thus we conclude that the ZPL linewidth, too, is affected by strain in the diamond lattice due to a modified electron–phonon coupling: at low temperatures the electron–phonon coupling is governed by a linear dynamical Jahn–Teller interaction between E-symmetric phonon modes (e_g phonons) and the SiV orbital states of E-symmetry [10, 38] as experimentally observed in measurements of ZPL linewidth versus temperature [10] and ZPL linewidth versus strain [6]. However, in a strongly strained environment, the D_{3d} symmetry of the SiV is

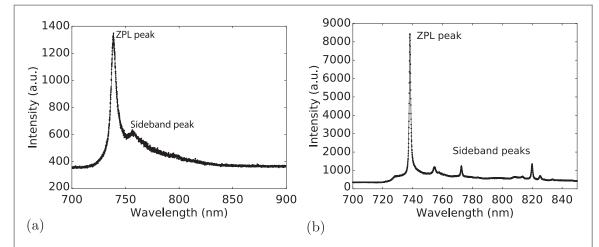


Figure 7. Representative photoluminescence spectra including phonon sidebands: (a) emitter of group V with a characteristic phonon sideband at \approx 42 meV; (b) emitter of group H with phonon sideband at \approx 36 meV (strain-shifted 42 meV sideband). For the narrow peaks in the sideband spectrum $g^{(2)}$ 760 nm an unambiguous attribution to phonon sidebands or weak emission of stress-shifted SiV centers is not possible.

lowered giving rise to an additional coupling of A-symmetry (a_{1g}) phonons which might further broaden the room-temperature ZPL emission as observed here.

To conclude, we are able to explain the distribution of ZPL center wavelengths in group V very consistently with theoretical predictions based on level shifts due to strain in the diamond lattice. On the other hand, we have to assume that group H is comprised of a new class of silicon-related defects, the structure of which is currently unclear.

3.2.3. Phonon sideband

From the literature it is known, that the SiV center in nanodiamond exhibits a large Debye–Waller factor of over 70% [18, 23], which is consistent with our measurements of emitter H1 and emitter V1. Nevertheless, sideband peaks are present in many SiV center PL emission spectra. The investigated emitters exhibit two different structures of sideband spectra: the spectra in group V commonly exhibit one strong sideband peak (figure 7(a)), whereas spectra in group H typically show several weaker sideband peaks (figure 7(b)).

About 70% of the group V PL spectra with one distinct sideband peak display a shift of the sideband peak from the ZPL between 37 and 43 meV. The range of line shifts for the prominent sideband peak coincides with a well-known feature at 42 meV, associated with SiV centers [55, 59], but also to a larger number of optically active defects [55]. The occurrence of this 42 meV sideband feature for a large number of defects and the absence of isotopic variations [36], favors an assignment as non-localized lattice vibration [60]. We furthermore observe that the dominant sideband peak shifts towards smaller distance from the ZPL for increasing ZPL center wavelength, i.e. increasing strain, see figure 8. We fit the data with a linear regression and find a slope of (-4 ± 1) meV nm⁻¹. The low phonon energy of the sideband feature and its shift with strain might arise from a local 'softening' of the crystal lattice in the vicinity of a defect [55]. Recent studies [38, 60] suggest that the 42 meV mode originates from a coupling of the ${}^{2}E_{u} - {}^{2}E_{g}$ optical transition to delocalized *e*_g phonons due to the dynamical Jahn–Teller effect in SiV centers. As the Jahn–Teller coupling varies with strain it is also expected that the resonance shifts accordingly.

In the spectra of group V, we do not observe a typical SiV center sideband feature at 64 meV, attributed to a local vibration of the silicon atom [36], frequently much stronger than the 42 meV sideband peak. A possible explanation is, that the lattice mode at 37–43 meV is so strong that the local vibrational mode at 64 meV cannot be separated from the tail of the lattice mode.

In group H we observe many spectra which exhibit several peaks within the spectral range of our detection range between 710 and 900 nm. The challenge arises to unequivocally distinguish between peaks stemming from a phonon sideband and peaks stemming from shifted, less intense SiV center ZPLs. We see, however, a tendency of peaks to accumulate at shifts of around 43, 64, 150, 175 meV. These findings are consistent with sideband shifts reported in [23, 55, 61].

3.3. Photon correlation measurements

The investigated individual SiV centers exhibit count rates of a few thousand to a few 100 000 counts per second (cps). We carried out measurements of the photon statistics and found that about 3% of luminescent nanodiamonds contain single color centers. Our measurements further reveal that the probability of finding a single emitter does not correlate in any way with the center wavelength or the linewidth of the ZPL. We found

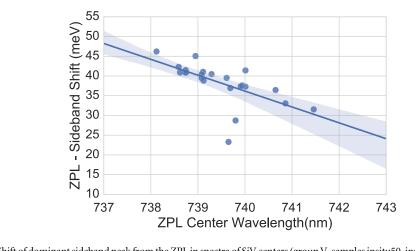


Figure 8. Shift of dominant sideband peak from the ZPL in spectra of SiV centers (group V, samples insitu50, insitu70, insitu100) versus ZPL center wavelength. The linear fit shows that the shift decreases with increasing ZPL center wavelength, i.e. with increasing strain and exhibits a slope of (-4 ± 1) meV nm⁻¹. The shaded area is the 95% confidence interval.

several single SiV centers with an antibunching dip down to about 0.2 and attribute the residual $g^{(2)}(0)$ value to background fluorescence from the diamond host. For the nanodiamonds used in our investigations an independent background measurement without simultaneously measuring SiV center PL is infeasible, because the laser spot size is bigger than the nanodiamond. Therefore, the background is estimated from the sideband of SiV center spectra. The measured lifetimes of the single SiV centers are in the range of about 1 ns to 9 ns in accordance with previous research [8, 18, 55].

Several nanodiamond PL spectra contain multiple narrow distinct peaks at different wavelengths. This circumstance is attributed to nanodiamonds containing more than one SiV center, each of which is subject to a different ZPL wavelength shift. We choose narrow bandpass filters to perform independent measurements of each individual peak of such a spectrum. As a result it is possible to measure $g^{(2)}(0)$ values below 0.5 for each of these narrow peaks. Hence the individual peaks are identified as single emitters with a different ZPL center wavelength.

We do not see a systematic difference regarding the photon autocorrelation functions of group H and group V, both reach similar $g^{(2)}(0)$ values. Also, the timescales of the excited state lifetimes coincide.

3.4. Photostability

As mentioned in the previous section, the single photon count rates observed from the investigated SiV centers varies strongly between a few thousand to a few 100 000 cps. To investigate the photostability, the luminescence time trajectory of the emitters which exhibit a dip in the photon correlation function with $g^{(2)}(0) < 0.5$ is evaluated. It is found that some of the observed emitters exhibit fluorescence intermittence, also called blinking (figure 9). Blinking is attributed to temporal ionization of the color center during optical excitation, forming a charge state which is optically inactive or emitting outside the detection window [21, 29, 62]. Therefore the emitters change between states of higher and lower emission, i.e. brighter and darker states, called blinking levels.

The photon count time trace of emitter emitter H1 is shown in figure 9. In the overview picture (figure 9(a)), a few blinking dips can be seen with time intervals of up to a few minutes. The fact that the count rate never drops down to the dark count rate lets us assume, that there are at least two SiV centers present, one exhibiting fluorescence intermittence and one exhibiting a stable emission or independent blinking. When zooming in, shorter time intervals are observable (figure 9(b)). The time intervals range from a few tens of milliseconds up to a few seconds with a few outliers exhibiting very long time intervals up to a few hundred seconds.

The bright and dark times exhibit different probability distribution functions and with that, different characteristic time constants. In figure 10 the time intervals of the emitter are shown as small vertical dashed red lines and solid blue lines for the bright and dark state respectively. To provide a smooth representation of the data, the blue and red dashed curves display kernel density estimations of the distribution of time intervals of the dark and bright states, respectively. This implies that every data point is represented with a Gaussian function and the resulting functions are added up to model the whole data. The red solid line is an exponential fit to the raw data distribution of time intervals in the bright state. The high *p*-value of 0.92 confirms the goodness of the fit. The median time interval in the bright state obtained by the exponential fit amounts to 0.09 s. While other literature on solid state quantum emitters reports an exponential probability distribution for both time intervals in bright and dark states [63, 64], we found a log-normal probability distribution for the time interval in the dark

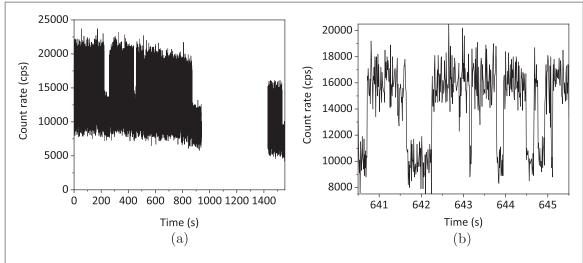


Figure 9. (a) Time trace of the single emitter H1, which exhibits strong blinking. The variation of the count rate in the upper state is attributed to a drift of the setup. (b) Detail of the time trace of the same emitter.

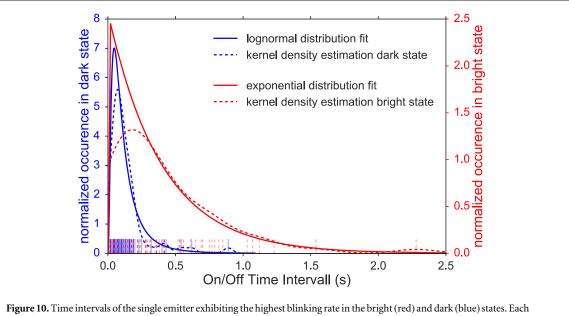


Figure 10. Time intervals of the single emitter exhibiting the highest blinking rate in the bright (red) and dark (blue) states. Each rising/falling edge of the count rate was individually read out from the time trace. On the horizontal axis small vertical lines represent the individual data points of the bright/dark time intervals. To provide a smooth representation of the data, the blue and red dashed curves represent kernel density estimations of the distribution of time intervals of the dark and bright states, respectively. The *y*-axis is scaled to the normalized kernel density estimate. Note that due to the boundary effect close to zero bright/dark times (x = 0) the kernel density estimators underestimate the true data distribution in this range. The red (blue) solid line is an exponential (log-normal) fit to the raw data of the bright (dark) state time intervals. These fitting functions were chosen because they provide the best agreement with the data using a Kolomogorov–Smirnov test with respect to other functions (*p*-values: bright state (red) 0.92, dark state (blue) 0.77).

state (solid blue line in figure 10). A Kolomogorov–Smirnov test yields a *p*-value of 0.77 for the log-normal fit and is by far the best model to describe the data distribution. For comparison, the *p*-value of an exponential fit amounts to 0.36. The median time interval in the dark state obtained by the log-normal fit is determined as 0.10 s, therefore being close to the median time interval in the bright state. Very long time intervals are not shown in the plot for better visualization of the small timescales, however these long time intervals are included in the fit. The longest measured time interval amounted to 41.14 s and occurred in the dark state. Measurements of SiV center blinking time intervals in [21, 29] report time intervals between about 0.03–1 s, and 0.1–2 min, respectively. These findings are in good agreement with our measurements. We do not identify a correlation between the count rate of a blinking state and its temporal duration. However, a correlation between the position in the bimodal distribution and blinking is established: all but one emitters in group H exhibit blinking, where only one of the emitters in group V exhibits blinking (figure 3). This dependency suggests that emitters in group H, i.e. silicon-related defects of currently unknown structure, are more likely to exhibit blinking. As blinking is typically linked to temporary loss of photo-excited charges or the modification of the charge environment [21] we tentatively explain the observed blinking as a manifestation of the local crystal disorder due to dislocations and impurities which act as a trap for charges and therefore influence the emitter's fluorescence state [63]. The assumption that dislocations and impurities are responsible for blinking emitters is in agreement with our findings reported in section 3.1. Regarding the time interval distributions, e.g. research of blinking rhodamine molecules confirmed power law distributed bright state times and log-normal distributed dark state times [65]. Log-normal distributions are typically explained by a Gaussian distribution of activation barriers of the electron transfer to trap states in the surrounding material [66] which hints towards a recapture of the electron via multi-phonon relaxation channels.

4. Conclusion

In conclusion, in this work we report on a strongly inhomogeneous distribution of SiV center spectra in nanodiamonds produced by wet milling from polycrystalline diamond films. We find that the zero phonon lines of the emission spectra can be grouped into two clusters: Group H consists of ZPLs exhibiting a narrow linewidth from below 1 nm up to 4 nm and a broad distribution of center wavelength between 715 and 835 nm. Compared to that, group V comprises ZPLs with a broad linewidth between just below 5 and 17 nm and center wavelength ranging from 730 to 742 nm. Based on *ab initio* DFT calculations we show that both the observed blue-shifts and red-shifts of the ZPLs of group V (as compared to an ideal, unstrained SiV center) are consistently explained by strain in the diamond lattice. Further, we suggest that group H might be comprised of fluorescent defects containing silicon, the structure of which is currently unclear. The broad distribution of emission wavelengths found here covers all earlier results on spectroscopy of SiV centers but considerably extends the range of known emission wavelengths. It further suggests that some single photon emitters in the 715–835 nm range, previously identified as Cr-, Ni- or Ni/Si-related (see e.g. [67]), could indeed exist of strained or perturbed SiV centers.

Whereas single photon emission could be demonstrated for SiV centers of both clusters, we show that the two clusters of SiV centers show different spectroscopic features: for the phonon sideband spectra we find in group V one prominent peak at a shift of 42 meV, which corresponds to a well-known feature assigned to non-localized lattice vibrations [55, 59]. In group H we see an accumulation of peaks, at around 43, 64, 150, 175 meV, which are consistent with sideband peaks reported in [23, 30, 55]. Investigating the time trace of the SiV center PL, we found that predominantly SiV centers with narrowband emission (group H) exhibit fluorescence intermittence with on/off times between several microseconds up to 41 s. Furthermore, we see an exponential distribution of bright time intervals and a log-normal distribution of dark time intervals, consistent with research on single molecules [65].

In summary, whereas SiV centers produced by ion implantation in high quality bulk diamond material show very reproducible spectral properties, SiV centers in nanodiamonds produced by milling techniques or CVD growth feature strongly varying optical spectra. This, on one hand, limits their applicability for quantum information tasks requiring indistinguishable emitters and, on the other hand, demonstrates the need for development of low-strain, nanometer-sized diamond material with low defect density.

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