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Citation: Journal of Vacuum Science & Technology A 34, 043201 (2016); doi: 10.1116/1.4948530
View online: http://dx.doi.org/10.1116/1.4948530
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Published by the AVS: Science & Technology of Materials, Interfaces, and Processing

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Nitridation of the GaAs (001) surface using atomic nitrogen
Planar regions of GaAs (001) prepared by Ga droplet motion

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(Received 4 November 2015; accepted 21 April 2016; published 10 May 2016)

The authors describe a simple method for obtaining planar regions of GaAs (001) suitable for surface science studies. The technique, which requires no buffer layer growth, atomic hydrogen source, or the introduction of As flux, employs controllable Ga droplet motion to create planar trail regions during Langmuir evaporation. Low-energy electron microscopy/diffraction techniques are applied to monitor the droplet motion and characterize the morphology and the surface reconstruction. It is found that the planar regions exhibit atomic flatness at the level of a high-quality buffer layer. © 2016 Author(s). All article content, except where otherwise noted, is licensed under a Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/). [http://dx.doi.org/10.1116/1.4948530]

I. INTRODUCTION

GaAs surfaces are of appreciable scientific and technological importance and have been actively studied over the years.1–3 The preparation of clean, atomically flat surfaces is paramount for fundamental studies of surface energetics, quantum structure formation, and epitaxial growth.1–3 With the emergence of in situ characterization techniques, such as transmission4 and low energy electron microscopy (LEEM),5,6 it is desirable to establish a simple means of creating planar surfaces for fundamental studies which do not demand a high As overpressure and which are compatible with the deposition geometry limitations imposed by electron optics. Here, we describe a simple method of smoothing GaAs (001) which utilizes the motion of Ga droplets and does not require material deposition.

The conventional approach to preparing planar GaAs (001) surfaces involves high temperature (~600°C) thermal desorption of the native oxide followed by the growth of thick (>0.5 μm) GaAs buffer layers and surface annealing under As flux.1–3 This approach cannot be utilized if buffer layer growth is not possible or if the introduction of As flux is undesirable. The alternative approach of atomic hydrogen (AH) cleaning7 offers a number of advantages, but annealing under As flux followed by subsequent buffer layer growth is ideally required to optimize the surfaces.8 Depending on the system design and geometry, it may not be possible or convenient to introduce such an AH source or indeed such equipment might not be available. Our method to create planar regions using Ga droplet motion involves no deposition or buffer layer growth and is compatible with all systems equipped with a heating stage.

II. EXPERIMENT

Experiments were performed in an ultrahigh-vacuum (UHV) LEEM.9 Temperatures T were measured using an infrared pyrometer, which incorporated a correction due to the T-dependence of the surface emissivity10 and was calibrated to the congruent evaporation temperature of GaAs given by $T_c = 625$ °C.11,12 We degassed a (001)-oriented undoped GaAs sample at 300°C for 24 h. This was followed by high temperature flashing up to 600°C and annealing at 580°C for 2 h to remove the surface oxide. This procedure necessarily creates a rough surface [Fig. 1(a)].

Our method to prepare droplets is based on the thermodynamics of Ga droplets on GaAs (001). During congruent evaporation ($T < T_c$), Ga and As evaporate at equal rates from the surface and the GaAs surface chemical potential $\mu_{Ga}$ attains a steady-state value. With increasing temperature, $\mu_{Ga}$ will increase to make As and Ga evaporation rates equal but eventually reaches the Ga liquidus value $\mu_L$ which defines the upper limit $T = T_c$ for congruent evaporation. Above $T_c$, $\mu_{Ga} > \mu_L$ so that excess Ga can collect as droplets which are assumed to remain close to liquidus composition.

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in equilibrium with the GaAs substrate. The droplets therefore act as sinks for surface Ga adatoms which pins $\mu_{Ga}$ close to $\mu_L$. This prevents the increase in $\mu_{Ga}$ so that above $T_c$, As evaporates more rapidly than Ga and the Ga droplets grow.

III. RESULTS AND DISCUSSION

Annealing the surface shown in Fig. 1(a) at 655 °C for 10 min creates four droplets in the field of view [Fig. 1(b)]. In mirror electron microscopy (MEM), the droplets appear as dark disks, somewhat larger than the actual droplet size, enclosed by a bright ring. Further droplet nucleation is prevented by rapidly reducing the substrate temperature to 640 °C [Fig. 1(c)]. Annealing at this temperature causes the droplets to grow, but their number remains constant. By adjusting the temperature in this way, it is possible to control the droplet density and size to optimize the smoothing procedure.

The droplets in Fig. 1(c), at liquidus composition, are out of equilibrium with the surrounding surface and this produces a force on the droplet proportional to its diameter. When the droplets attain a large enough size, this force overcomes the pinning force due to droplet etching and the droplets move along $\pm [110]$ directions, leaving behind smooth trails [Fig. 1(d)].

Figure 2(a) shows an LEEM image of the trail which consists of planar terrace regions separated by monolayer surface steps. The step density is smaller than 15, and the surface morphology is comparable to that obtained by thick buffer layer growth. The micro-low-energy electron diffraction ($\mu$LEED) pattern from the trail contained in Fig. 2(b) corresponds to a $c(8 \times 2)$ reconstruction. This is in contrast to the $\mu$LEED pattern of the region outside the trail which, in addition

![Fig. 1. MEM images of Ga droplet formation and dynamics.](image)

![Fig. 2.](image)
to c(2 × 8) spots, also contains additional spots which move with electron beam energy and can be linked to facets present in the surface roughness. The general roughness and associated facets are related to thermal oxide desorption followed by long-time surface annealing at high temperature. It can be seen that running droplets are effective tools for removing this roughness and creating atomically flat, planar regions for surface science studies. Although it is known that buffer layers can additionally bury low-level chemical impurities, it is presently unclear as to whether running droplets can provide similar effects by surface absorption.

By a judicious choice of sample heating to control droplet nucleation and motion, it is possible to approach a smooth surface coverage of 90%. Droplets can then be removed by annealing below $T_c$. Even higher percentages of smooth surface can be obtained by droplet nucleation and motion, but this generally involves the formation of large (>10 μm) droplets which are impractical to remove by annealing below $T_c$ due to the protracted times involved. Nevertheless, such large areas of smooth surface open up the possibility of using chemical spectroscopies, or standard surface electron diffraction techniques. Smaller trail regions, as in Fig. 2, for example, are suitable for the study by local and semilocal probes such as the scanning tunneling microscopy, scanning electron microscopy, and LEEM. Moreover, droplet motion and control has been observed across a variety of III–V material systems, which may lead to broader applications of the method discussed here.

IV. SUMMARY AND CONCLUSIONS

In summary, we have described a simple method for obtaining planar regions of GaAs (001) suitable for surface science studies. The method requires no buffer layer growth, AH source, or the introduction of As flux. If real space imaging is available, the process can be fine-tuned by controlling the temperature for droplet formation and motion. However, \textit{ex situ} analysis of test surfaces annealed at different temperatures should allow this method to be routinely used in any UHV system.

ACKNOWLEDGMENTS

C.Z. acknowledges support from Australian Research Council (ARC) and D.E.J. acknowledges support from a Marie Curie International Incoming Fellowship.